FORMAL SYNTHESIS OF (±)-IPALBIDINE

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l-Aza-3-(4-methoxyphenyl)bicyclo[4,3,0]nonan-4-one, key intermediate to (\pm) -ipalbidine, was conveniently synthesized by intra-molecular cyclization of the N-formylketone which was obtained via [3 + 2] cycloaddition reaction of the cyclic nitrone.

Ipalbidine $\underline{1}$, the aglycone of ipalbine $\underline{2}$ and ipomine $\underline{3}$ isolated from seeds of *Ipomoea alba* $\underline{L}.^{1)}$ and *Ipomoea muricata* \underline{J} acq., $\underline{J}^{2},3)$ respectively, is the only instance of an indolizidine alkaloid with methyl substitution on the indolizidine nucleous occurring in nature. Among reported syntheses of ipalbidine, \underline{J}^{5-9}

three syntheses have been concerned with the synthetic sequence utilizing the bicyclic ketone $\underline{12}$ as the key intermediate, which have been prepared by Dieckmann condensation⁵⁾ and intramolecular cyclization of an enamine⁷⁾ and vinylogous urethane.⁹⁾ In this report we describe a new, convenient procedure for the preparation of this intermediate. The key feature of the synthetic method involves highly selective 1,3-dipolar cycloaddition¹⁰⁾ in the synthesis of the β -amino alcohol $\underline{7}$, which was converted in subsequent steps with final intramolecular ring closure into the bicyclic ketone intermediate 12.

1: R = H

2: $R = \beta-D-glucopyranosyl$

3: R = 6-O-p-coumaroyl-β-D-glucopyranosyl

The reaction of 1-pyrroline 1-oxide $\underline{4}$ with p-allylanisole $\underline{5}$ in refluxing toluene produced the isoxazolidine $\underline{6}$ (oil) in 70% yield as the sole product. The high regionelectivity in the desired sense observed in this reaction is most explicable on the basis of the preference of dipole LUMO control 11,12) with nonconjugated electron-rich dipolarophile. Subsequent N-O bond cleavage of $\underline{6}$ by hydrogenation (10% Pd/C) gave the β -amino alcohol $\underline{7}$ (mp 97 °C) in 82% yield. Thus functionalization required for the target was simply achieved at this early stage.

The amino alcohol $\underline{7}$ thus obtained was heated with formic acid in toluene to furnish the O,N-diformate $\underline{8}$ together with the N-formylcarbinol $\underline{9}$. For selective hydrolysis of the O-formate group, this mixture was subsequently exposed to ammonia in methanol to give the desired N-formylcarbinol $\underline{9}$ (oil) in 69% yield from $\underline{7}$. Collins oxidation (CH₂Cl₂, room temp) of $\underline{9}$ yielded the N-formylketone $\underline{10}$ (oil) in 72% yield. The cyclization of $\underline{10}$ to $\underline{11}$ by aldol reaction is unusual since the amide C=0 is normally unreactive partner as the electrophile in this

reaction. After several unsuccessful attemts, however, cyclization was attained when heated with aluminum t-butoxide $^{14)}$ in xylene, providing the bicyclic enaminone 11 (mp 121-122 °C) in 36% yield. Eventually selective reduction of the olefinic double bond of 11 was achieved by using lithium in liquid ammonia to afford the bicyclic ketone 12 (mp 109-110 °C) in 54% yield which had mp and spectral data identical to those reported in the literature. 9) Since the compound 12 has already been converted into (\pm) -ipalbidine 1 by two steps, 5,7) our preparation of this compound constitutes a formal total synthesis of racemic 1.

Ar
$$\frac{10}{12}$$

Ar $\frac{H}{H}$

Ar $\frac{H}{H}$

Ar $\frac{H}{H}$
 $\frac{10}{12}$

Ar $\frac{H}{H}$
 $\frac{Ar}{H}$
 $\frac{10}{H}$
 $\frac{11}{Ar}$
 $\frac{11}{H}$
 $\frac{11}{Ar}$
 $\frac{12}{H}$
 $\frac{12}{H}$
 $\frac{Ar}{H}$
 $\frac{10}{H}$
 $\frac{11}{H}$
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